

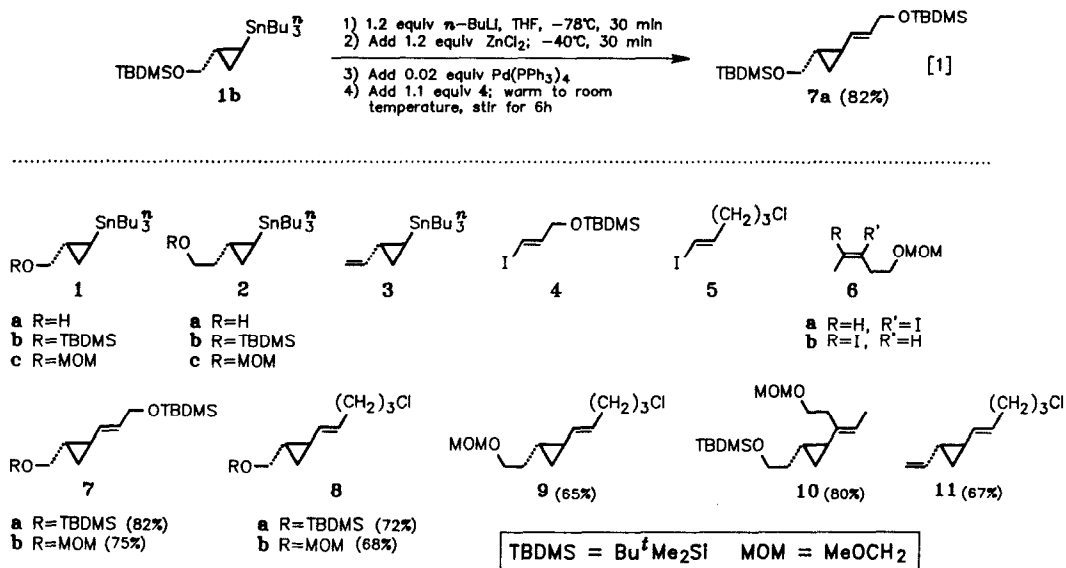
**SYNTHESIS OF VINYL-CYCLOPROPANES VIA PALLADIUM-CATALYZED COUPLING OF CYCLOPROPYLZINC HALIDES WITH VINYL IODIDES. TOTAL SYNTHESSES OF ( $\pm$ )-PREZIZANOL AND ( $\pm$ )-PREZIZAENE**

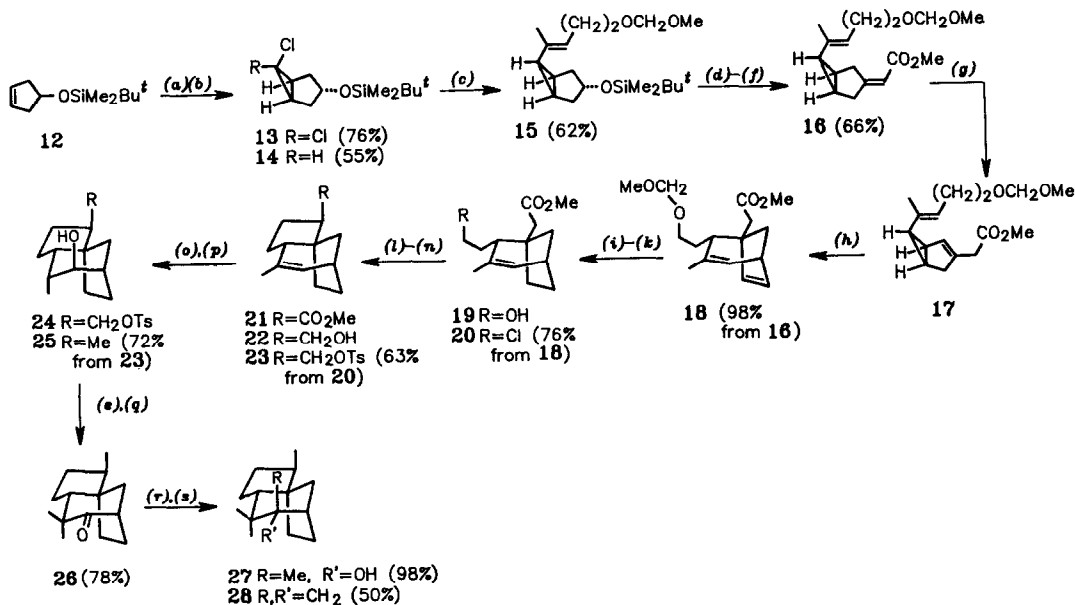
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**ABSTRACT:** Palladium(0)-catalyzed coupling of cyclopropylzinc halides, readily prepared from the corresponding cyclopropyl(tri-*n*-butyl)stannanes, with vinyl iodides provides good yields of functionalized vinylcyclopropanes. The coupling reaction is used as a key step in total syntheses of the sesquiterpenoids ( $\pm$ )-prezizanol and ( $\pm$ )-prezizaene.

During the past 10 - 15 years, the coupling of organometallic reagents with structurally diverse organic halides has become a versatile and increasingly important method of forming carbon-carbon bonds.<sup>1</sup> For example, a wide variety of functionalized organic compounds are readily available by palladium- and nickel-catalyzed coupling of various organometallic species with aryl, alkenyl, and alkynyl halides.<sup>1</sup> Furthermore, also in the recent past, investigations into the use of functionalized cyclopropanes in organic synthesis have proven to be very fruitful.<sup>2</sup> In this regard, the synthetic uses of thermal rearrangements of vinyl- and 1,2-divinylcyclopropane systems have been particularly noteworthy. In this Letter, we describe a new method of preparing functionalized vinylcyclopropanes by palladium(0)-catalyzed coupling of cyclopropylzinc halides with vinyl iodides and illustrate the use of this reaction as a key step in total syntheses of the sesquiterpenoids ( $\pm$ )-prezizanol and ( $\pm$ )-prezizaene.

The reaction conditions employed for the overall coupling of the tri-*n*-butylstannylcyclopropane **1b**<sup>3,4</sup> with the vinyl iodide **4**<sup>7</sup> are given in Eq. [1].<sup>9</sup> In similar fashion, various combinations of tri-*n*-butylstannylcyclopropanes and vinyl iodides (**1c**<sup>4</sup> + **4**,<sup>7</sup> **1b**<sup>4</sup> + **5**,<sup>7</sup> **1c**<sup>4</sup> + **5**,<sup>7</sup> **2c**<sup>4</sup> + **5**,<sup>7</sup> **2b**<sup>4</sup> + **6a**,<sup>7</sup> **3**<sup>4</sup> + **5**<sup>7</sup>) afforded the substituted cyclopropanes **7b**, **8a,b**, and **9 - 11**, respectively, in the yields indicated in Chart 1. In each case, the reaction was quite clean and the product was purified by chromatography (silica gel) and distillation. In the preparation of compound **11**, ZnBr<sub>2</sub> was used in the place of ZnCl<sub>2</sub> and, since the product **11** was difficult to separate from the vinyl iodide **5**, a slight deficiency (0.95 equiv) of the latter material was employed. It can be seen that the reaction tolerates a number of different functional groups, the vinyl iodide coupling partner may possess an alkyl substituent on the carbon bearing the iodine (**2b** + **6a** → **10**), and the method is amenable to preparing 1,2-divinyl-(e.g. **11**) as well as monovinylcyclopropanes.





**Scheme 1.** (a) CHCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, NaOH, H<sub>2</sub>O, PhCH<sub>2</sub>Et<sub>3</sub>NCl; (b) *t*-BuLi, THF-Et<sub>2</sub>O-pentane (3:1:1), -107 °C; HOAc, Et<sub>2</sub>O, -107 °C; (c) (4,4'-di-*tert*-butylbiphenyl)<sup>+</sup>Li<sup>+</sup>, THF, -78 °C; ZnCl<sub>2</sub>, THF, 0 °C; 6b, Pd(PPh<sub>3</sub>)<sub>4</sub>, THF, reflux; (d) *n*-Bu<sub>4</sub>NF, THF; (e) C<sub>5</sub>H<sub>5</sub>NCrO<sub>3</sub>HCl, NaOAc, CH<sub>2</sub>Cl<sub>2</sub>; (f) [(MeO)<sub>2</sub>POCHCO<sub>2</sub>Me]Li, THF; (g) *i*-Pr<sub>2</sub>NLi, THF, -78 °C; HOAc, Et<sub>2</sub>O, -78 °C; (h) 110 °C, 0.1 torr; (i) H<sub>2</sub>, (Ph<sub>3</sub>P)<sub>3</sub>RhBr, C<sub>6</sub>H<sub>6</sub>; (j) *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H, MeOH, reflux; (k) CCl<sub>4</sub>, Ph<sub>3</sub>P, Et<sub>3</sub>N, reflux; (l) *i*-Pr<sub>2</sub>NLi, THF, -78 °C; (m) LiAlH<sub>4</sub>, Et<sub>2</sub>O; (n) *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>Cl, 4-dimethylaminopyridine, CH<sub>2</sub>Cl<sub>2</sub>; (o) BH<sub>3</sub>·Me<sub>2</sub>S, THF; H<sub>2</sub>O<sub>2</sub>, NaOH; (p) LiEt<sub>3</sub>BH, THF; (q) KH, THF; MeI, THF, reflux; (r) MeLi, THF, -78 °C; (s) MeSO<sub>2</sub>Cl, *i*-Pr<sub>2</sub>NEt, CH<sub>2</sub>Cl<sub>2</sub>.

(-10%) of the epimeric ester.<sup>16</sup> Reduction afforded the corresponding alcohols (22 + epimer), which were readily separated by chromatography of the corresponding *tert*-butyldimethylsilyl ethers.

A standard sequence of reactions was employed to convert the alcohol 22 into the racemic ketone 26, which was spectrally identical with (-)-26 previously synthesized by Vettel and Coates.<sup>12,17</sup> Treatment of 26 with MeLi provided (±)-prezizanol (27), which, upon dehydration, gave (±)-prezizaene (28). The latter materials gave IR and <sup>1</sup>H NMR spectra identical with those of (-)-27<sup>17</sup> and (-)-28,<sup>17</sup> respectively.

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  - All isolable substances reported herein exhibited spectra in accord with structural assignments and gave satisfactory high resolution mass spectrometric molecular mass measurements.
  - Compounds 1b,c and 2b,c were prepared by reaction of 1a and 2a with *t*-BuMe<sub>2</sub>SiCl (imidazole, Me<sub>2</sub>NCHO) or MeOCH<sub>2</sub>Cl (*i*-Pr<sub>2</sub>NEt, CH<sub>2</sub>Cl<sub>2</sub>). The alcohols 1a and 2a were acquired by cyclopropanation (CH<sub>2</sub>I<sub>2</sub>, Et<sub>2</sub>Zn, PhMe, 40 °C)<sup>5</sup> of the corresponding alkenes, which were prepared by hydrostannylation (1.3 equiv *n*-Bu<sub>3</sub>SnH, AIBN, 80 °C)<sup>6</sup> of 2-propyn-1-ol and 3-propyn-1-ol. Compound 3 was obtained from 1a by oxidation (pyridinium chlorochromate, NaOAc, CH<sub>2</sub>Cl<sub>2</sub>), followed by Wittig olefination (Ph<sub>3</sub>P=CH<sub>2</sub>, THF).
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  - The vinyl iodides 4 - 6 were prepared by hydrozirconation-iodination<sup>8</sup> of the corresponding alkynes. In the case of MeC≡C(CH<sub>2</sub>)<sub>2</sub>OCH<sub>2</sub>OMe, this reaction produced a 1:1 mixture of 6a and 6b, which were separated by chromatography on silica gel. The isolated yields were: 6a, 38%; 6b, 36%.
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  - Compound 17 rearranged slowly to 18 at room temperature. Interestingly, thermolysis of the *exo* isomer of 17 at various temperatures (155 - 220 °C, benzene, sealed tube) produced a plethora of products from which 18 could be isolated in only poor yield (~25%).
  - A separate experiment showed that this epimeric ester was formed by epimerization of the primary product 21. Thus, treatment of 20 with a deficiency of *i*-Pr<sub>2</sub>NLi gave exclusively 21, accompanied by some starting material 20.
  - We are very grateful to Professor Coates for copies of the IR and <sup>1</sup>H NMR spectra of (-)-26, (-)-prezizanol, and (-)-prezizaene.

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