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SYNTHESIS OF VINYLCYCLOPROPANES VIA PALLADIUM-CATALYZED COUPLING OF CYCLOPROPYLZINC HALIDES WITH VINYL IODIDES. TOTAL SYNTHESES OF (±)-PREZIZANOL AND (±)-PREZIZAENE

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ABSTRACT: Palladium(O)-catalyzed coupling of cyclopropylzinc halides, readily prepared from the corresponding $cyclopropy(tri-p-butyl)$ stannanes, with vinyl iodides provides good yields of functionalized vinylcyclopropanes. The coupling reaction is used as a key step in total syntheses of the sesquiterpenoids (t) -prezizanol and (t) -prezizaene.

During the past 10 - 15 years, the coupling of organometallic reagents with structurally diverse organic halides has become a versatile and increasingly important method of forming carbon-carbon bonds. 1 For example, a wide variety of functionalized organic compounds are readily available by palladium- and nickel-catalyzed coupling of various organometallic species with aryl, alkenyl, and alkynyl halides. 1 Furthermore, also in the recent past, investigations into the use of functionalized cyclopropanes in organic synthesis have proven to be very fruitful.² In this regard, the synthetic uses of thermal rearrangements of vinyl- and 1,2-divinylcyclopropane systems have been particularly noteworthy. In this Letter, we describe a new method of preparing functionalized vinylcyclopropanes by palladium(O)-catalyzed coupling of cyclopropylzinc halides with vinyl iodides and illustrate the use of this reaction as a key step in total syntheses of the sesquiterpenoids (t) -prezizanol and (t) -prezizaene.

The reaction conditions employed for the overall coupling of the tri-n-butylstannylcyclopropane $1b^{3,4}$ with the vinyl iodide 4^{7} are given in Eq. [1].⁹ In similar fashion, various combinations of tri-n-butylstannylcyclopropanes and vinyl iodides ($1c^4$ + 4 , $\frac{7}{1}$ lb⁴ + 5.⁷ $1c^4$ + 5.⁷ $2c^4$ + 5.⁷ $2b^4$ + 6a.⁷ 3^4 + 5⁷) afforded the substituted cyclopropanes 7b, Sa,b, and 9-11, respectively, in the yields indicated in Chart 1. In each case, the reaction was quite clean and the product was purified by chromatography (silica gel) and distillation. In the preparation of compound 11, ZnBr₂ was used in the place of ZnCl₂ and, since the product 11 was difficult to separate from the vinyl iodide 5, a slight deficiency (0.95 equiv) of the latter material was employed. It can be seen that the reaction tolerates a number of different functional groups, the vinyl iodide coupling partner may possess an alkyl substituent on the carbon bearing the iodine (2b + $6a \rightarrow 10$), and the method is amenable to preparing 1,2-divinyl-(e.g. 11) as well as monovinylcyclopropanes.

The sesquiterpenoid prezizaene (enantiomer of 28) was first isolated from vetiver oil of Reunion origin, 10 while (-)-prezizaene (28) and the related alcohol (-)-prezizanol (27) were obtained from <u>Eremophila georgei</u>.¹¹ Total syntheses of (-)-27 and (-)-28 have been reported.¹² Outlined in Scheme 1 are our total syntheses of (\pm)-27 and (\pm)-28 via a route in which the Pd(O)-catalyzed coupling of a cyclopropylzinc chloride with the vinyl iodide $6b^7$ played a key role.

Reaction of the silyl ether 12^{13} with dichlorocarbene gave primarily 13, accompanied by a small amount of the corresponding cis isomer. Slow addition of a THF solution of 13 to a cold solution of t -BuLi in 3:1:1 THF-Et₂0-pentane, followed by protonation of the intermediate and subsequent chromatographic separation of the mixture of chlorocyclopropanes, provided the desired endo isomer 14 in 55% yield. Treatment of 14 with lithium $4,4'$ -di-tert-butylbiphenylide, 14 conversion of the resultant cyclopropyl-lithium reagent into the corresponding organosinc chloride, and $Pd(PPh_3)_4$ -catalyzed coupling of the latter species with the iodide 6b afforded 15, which was converted efficiently via a 3-step sequence into the α, β -unsaturated ester 16. Deconjugation of 16 gave 17, which, upon distillation at 110 "C, underwent clean Cope rearrangement¹⁵ to give the bicyclic diene 18.

Chemoselective hydrogenation of 18, followed by treatment of the resultant alkene with p-MeC₆H₄SO₃H-MeOH gave an inseparable equilibrium (3:2) mixture of the alcohol ester 19 and the corresponding lactone. Reaction of this mixture with $Ph_3P\text{-}CCl_{\Delta}\text{-}Et_3N$, followed by chromatography of the product mixture, produced the desired chloride 20 (48%) along with the lactone (41%). The latter material could be recycled (p-MeC₆H₄SO₃H, MeOH; $Ph_3P-CC1_4-Et_3N$; separation) and, in this manner, a 76% overall yield of the chloride 20 from 18 was obtained after two recycles.

Examination of molecular models leads one to predict that intramolecular alkylation of the ester chloride 20 would provide mainly or exclusively the tricyclic ester 21. In the event, treatment of 20 with i -Pr₂NLi in THF afforded 21, accompanied by a small amount

Scheme 1. (a) CHCl₃, CH₂Cl₂, NaOH, H₂O, PhCH₂Et₃NCl; (b) E -BuLi, THF-Et₂O-pentane $(3:1:1)$, -107 °C; HOAc, Et₂0, -107 °C; (c) $(4, 4'$ -di-tert-butylbiphenyl)⁺Li⁺, THF, -78 °C; $ZnCl_2$, THF, 0 °C; 6b, Pd(PPh₃)₄, THF, reflux; (d) p_B-Nu_ANF , THF; (e) C₅H₅NCrO₃HC1, NaOAc, CH_2Cl_2 ; (f) $[(Me0)_2PCCHCO_2ME]Li$, THF; (g) $i Pr_2NLi$, THF, -78 °C; HOAc, Et₂0, -78 °C; (h) 110 °C, 0.1 torr; (i) H₂, (Ph₃P)₃RhBr, C₆H₆; (j) p-MeC₆H₄SO₃H, MeOH, reflux; (k) CCl₄, Ph₃P, Et₃N, reflux; (1) i-Pr₂NLi, THF, -78 °C; (m) LiAlH₄, Et₂O; (n) p-MeC₆H₄SO₂Cl, 4-dimethylaminopyridine, CH₂Cl₂; (o) BH₃'Me₂S, THF; H₂O₂, NaOH; (p) LiEt₃BH, THF; (q) KH, THF; MeI, THF, reflux; (r) MeLi, THF, -78 °C; (s) MeSO₂Cl, i-Pr₂NEt, CH₂Cl₂.

 $(-10*)$ of the epimeric ester.¹⁶ Reduction afforded the corresponding alcohols (22 + epimer), which were readily separated by chromatography of the corresponding tert-butyldimethylsilyl ethers.

A standard sequence of reactions was employed to convert the alcohol 22 into the racemic ketone 26, which was spectrally identical with (-)-26 previously synthesized by Vettel and Coates.^{12,17} Treatment of 26 with MeLi provided (\pm)-prezizanol (27), which, upon dehydration, gave (\pm) -prezizaene (28). The latter materials gave IR and 1 H NMR spectra identical with those of (-)-2 7^{17} and (-)-28, 17 respectively.

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- 3. All isolable substances reported herein exhibited spectra in accord with structural assignments and gave satisfactory high resolution mass spectrometric molecular mass measurements.
- 4. Compounds 1b,c and 2b,c were prepared by reaction of 1a and 2a with t-BuMe₂SiCl (imidazole, Me₂NCHO) or MeOCH₂Cl (i-Pr₂NEt, CH₂Cl₂). The alcohols la and 2a were acquired by cyclopropanation (CH₂I₂, Et₂Zn, PhMe, 40 °C)⁵ of the corresponding alkenes, which were prepared by hydrostannylation (1.3 equiv $p_B - B u_3 S nH$, AIBN, 80 °C)⁶ of 2-propyn-l-01 and 3-propyn-l-01. Compound 3 was obtained from la by oxidation (pyridinium chlorochromate, NaOAc, CH_2Cl_2), followed by Wittig olefination (Ph₃P=CH₂, THF).
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- 7. The vinyl iodides 4 6 were prepared by hydrozirconation-iodination^o of the corresponding alkynes. In the case of $MeC=C(H_2)_2OCH_2OMe$, this reaction produced a 1:l mixture of 6a and 6b, which were separated by chromatography on silica gel. The isolated yields were: 6a, 38%; 6b, 36%.
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- 15. Compound 17 rearranged slowly to 18 at room temperature. Interestingly, thermolysis of the $\underline{\text{exo}}$ isomer of 17 at various temperatures (155 - 220 °C, benzene, sealed tube) produced a plethora of products from which 18 could be isolated in only poor yield $(-25%)$.
- 16. A separate experiment showed that this epimeric ester was formed by epimerization of the primary product 21. Thus, treatment of 20 with a deficiency of i -Pr₂NLi gave exclusively 21, accompanied by some starting material 20.
- 17. We are very grateful to Professor Coates for copies of the IR and ¹H NMR spectra of (-)-26, (-)-prezizanol, and (-)-prezizaene.

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